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## Article 1

**Svetskaya N. V., Lukina Yu. S., Zaitsev A. S.**

### **Silicon structured hydroxyapatite cements for bone plastic surgery**

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**Keywords:** silicon structured calcium phosphates, tricalcium phosphate, hydroxyapatite, hydroxyapatite cement, porosity, durability

## Abstract

Silicon structured hydroxyapatite cements are obtained on the basis of silicon structured  $\alpha$ -tricalcium phosphates having the increased values of solubility. Influence of content of silicon in structure silicon structured  $\alpha$ -tricalcium phosphates and concentration of tempering water – sodium hydrophosphate on chemical and physicomachanical properties of the received cements is investigated. Silicon structured hydroxyapatite cements for application in bone plastic surgery are optimized.

## References

1. Carlisle E. Si: an essential element for the chick // *Science*. – 1972. – Vol. 178. – P. 619-621.
2. Schwarz K., Milne D. Growth promoting effects of Si in rats // *Nature*. – 1972. – Vol.239. – P. 333-334.
3. Voronkov M. G., Zelchan G. I., Lukevits E. Ya. Kremniy i zhizn' [Silicium and life]. – Riga. Znaniye, 1978. – 552 S.
4. Alexis M. Pietak, Joel W. Reid, Malcom J. Stott, Michael Sayer Silicon substitution in the calcium phosphate bioceramics // *Biomaterials* 2007; 28. – P. 4023-4032.
5. Portera A. E., Patela N., Skepperb J. N., Besta S. M., Bonfielda W. Comparison of in vivo dissolution processes in hydroxyapatite and silicon-substituted hydroxyapatite bioceramics // *Biomaterials* 2003; 24. – P. 4609-4620.
6. Karin A. Hinga, Peter A. Revellb, Nigel Smithc, Thomas Buckland Effect of silicon level on rate, quality and progression of bone healing within silicatesubstituted porous hydroxyapatite scaffolds // *Biomaterials* 2006; 27. – P. 5014-5026.
7. Mastrogiacomoa M., Papadimitropoulosb A., Cedolac A., Peyrind F., Giannonie P. Engineering of bone using bone marrow stromal cells and a silicon-stabilized tricalcium phosphate bioceramic: Evidence for a coupling between bone formation and scaffold resorption // *Biomaterials* 2007; 28. – P. 1376-1384.

## Article 2

**Molchan N. V., Krivoborodov Yu. R., Fertikov V. I.**

### **Interatomic interactions in binary compounds of calcium**

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**Keywords:** concentration of electrons, density, structure

## Abstract

Chemical interactions and phase transformations are traditionally characterized by thermodynamic indicators and state diagram. Transformations of substances are followed by thermal effects and changes of volume. Density, important characteristic of substances, is result of two indicators: 1) the weight which is concentrated in atomic nuclei and 2) the volume which is formed by electron shells. Chemical processes are the reactions proceeding with formation of new connections. The act of chemical interaction consists in formation of new electronic (molecular) orbitals. The chemical bond between atoms is caused by overshoot of electronic clouds. Transformations of substances and formation of new structure is defined by interaction of electron shells of atoms and molecules. Thermal processes are

rather in detail considered in numerous works on chemical thermodynamics, and information on changes of volumes is not enough.

## References

1. Molchan N. V., Fertikov V. I. (2011) Compressibility of Substances and Dimensions of Atoms. *Material Science*, 6, 2-6. (in Russian).
2. Molchan N. V., Fertikov V. I. Determination of Concentration of Electrons for Description of the Structure of Materials, with Sulfides as an Example // *Journal of Materials Sciences and Applications*. 2015. V. 1. № 2. P. 38-44.
3. Molchan N. V., Fertikov V. I. Interrelation of Thermodynamic Parameters and Structural Characteristics, with Halides of Groups 1 and 2 Elements as an Example. 2016. *American Journal of Chemistry and Application*. Vol. 3, No. 5, pp. 28-32.
4. Molchan N. V., Fertikov V. I. Concentration of electrons as a structural characteristic of oxides, Technique and technology of silicates, 2016; 2: 8-13 (in Russian).
5. Molchan N., Krivoborodov Yu., Fertikov V. The interaction of water with oxides, forming hydroxides and crystal hydrates, Technique and technology of silicates, 2017; 1: 11-15 (in Russian).
6. Molchan N., Eliseev D., Fertikov V. Control of Nickel Alloy Structural Change by the Atomic Emission Spectroscopy Method. *American Journal of Analytical Chemistry*, 2016, vol. 7, no. 9, pp. 633-641.
7. Fertikov V., Seguru G. Assessment of Changes in Volume of Nickel Compounds Interacting with the Chemical Elements. *International Journal of Current Research*. 2017, vol 9, Issue, 08, pp. 56361-56364.
8. Seguru G., Fertikov V. Interaction of Elements in Binary Compounds of Hydrogen. 2017. *American Journal of Chemistry and Application*. 2017. Vol. 4, No. 6, Page: 59-62
9. Molchan N., Krivoborodov Yu., Fertikov V. The interaction of silicon with the chemical elements, forming with it a binary connection. Technique and technology of silicates, 2017; 4: 11-17 (in Russian).
10. International Centre for Diffraction Data. JCPDS PCPDFWIN, 2002; V. 2.03.
11. Key Properties of Inorganic, Organic and Element Organic Compounds, The New Reference Book for Chemists and Technologists, NPO «Professional», SaintPetersburg, 2007.
12. Lidin R., Andreeva L., Molochko L. 2006. *Konstantineorganicheskikh veshchestv. Spravochnik* [The constants of inorganic substances. Reference book], Drofa, Russia.

## Article 3

**Kozlova I. V., Nechaev K. V.**

### **Effect of thin-dispersible slag on cement properties with mineral additives**

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**Keywords:** fine-milled slag, granulometric composition, cement stone, strength, porosity, degree of hydration

## Abstract

The method of introduction of fine-milled blast-furnace granulated slag (DGS) into the cement composition as a result of dry mixing of the material with the additive is considered. Fine-milled slag was obtained in a vortex jet mill with the upper limit of grinding to 1 and 20 microns (slag 1, slag 2, respectively). The granulometric composition of slag 1 and 2 is determined. It was found that the predominant size in slag 1 is 0.5–1 μm; in slag 2 – 1–7 μm. Fine-milled slag (slag 1, slag 2) was introduced into the cement, containing in its composition as an active mineral additive 15% DGS (particle size – 40-60 microns), in an amount of 1, 3, 5% of the cement content. The aim of the study was to study the effect of the addition of fine slag on the construction, physical, mechanical, structural characteristics of cement with mineral additives. The terms of setting were studied, the normal density of the cement paste with the addition of fine slag was determined. Dependences of strength and porosity of cement stone on the time of hydration of samples are constructed. Strength of cements was determined by the national standard GOST 30744-2001. The porosity of cement stone was determined by the method of saturation of samples with inert liquid. It is established that the introduction of 3–5% slag 1 and 1–3% slag 2 provide increased strength of the cement stone during the first day of hardening on average, 37 to 44%; in the branded age of 26–30 %, reduction of porosity by 17–28%. It is shown that the introduction of a fine-milled slag additive compacts and strengthens the structure of cement stone. It was found that grinding DGS to a size of 1 μm (slag 1) is impractical, because the obtained results are comparable with the results of the introduction of slag 2 into the cement composition.

## References

1. Khigerovich M. I. Works of Russian scientists on the technology of building binders / Ed. B.G. Skramtaeva. M: IISS, 1948. 36 p (in Russian).

2. Fedorov P. A., Abdullin M. M., Potapova O. G., Stepanova M. Yu., Reshetnev G. V., Andrianova S. O. The study of physical and mechanical properties and durability of concrete and reinforced concrete. Contribution N. A. Beleyubsky into Russian concrete science. History of science and technology. 2017. № 6. pp. 66-74 (in Russian).
3. Benin A. V. Professor N. A. Beleyubsky and the «Golden Age» of the mechanical laboratory. Cement and its application. 2015. № 3. pp. 140-141.
4. Tsipursky I. L., Kokonova A. A., Danilova E. D., Kovchenko I. V., Rudenko M. I. Domain granulated slags in the production of multicomponent cement systems: production technology and application features. Transport facilities. 2018. Vol. 5. No. 1. pp. 1-8. DOI: 10.15862 / 18SATS118.
5. Ufimtsev V. M., Kapustin F. L. Thermal activation of the binding properties of mineral melts. Concrete Technologies. 2014. № 1 (90). pp. 19-21.
6. Gusev B. V., In I.L.S., Krivoborodov Yu. R. Activation of hardening of slag Portland cement // Concrete Technologies. 2012. № 7-8 (72-73). pp. 21-24.
7. Trofimov B. Ya., Kramar L. Ya., Shuldyakov K. V. The influence of the amount of slag in cement on the frost resistance of heavy concrete. Construction materials. 2013. № 9. pp. 96-101.
8. Kiel P. N., Kirsanova A. A., Kramar L. Ya., Trofimov B. Ya., Dobrovolsky I. P. Additives-accelerators of polyfunctional action for slag Portland cement. Bulletin of the South Ural State University. Series: Building and Architecture. 2014. V. 14. № 2. pp. 27-32.
9. Petrova T. M., Smirnova O. M., Frolov S. T. Properties of plasticized compositions of Portland cementblast furnace slag taking into account electro-surface phenomena. Bulletin of Civil Engineers. 2011. № 2 (27). pp. 118-123.
10. Zagorodnyuk L. Kh. Steel slag - low-power filler of composite materials. Dry building mixes. 2012. № 1. pp. 14-15.
11. Ufimtsev V.M., Korobeinikov L.A. Slags in the composition of concrete: new opportunities. Concrete Technologies. 2014. № 6 (95). pp. 50-53.
12. Samchenko S. V., Dudoladova T. G. Influence of the nature of crystallization of alumina slag on the properties of alumina cement. Technique and technology of silicates. 2004. V. 11. No. 1-2. P. 24.
13. Samchenko S. V., Lyutikova T. A., Kuznetsova T. V., Dudoladova T. G. Improving the properties of alumina cement and its application // Cement and its application. 2006. № 3. pp. 46-48.
14. Samchenko S. V., Zorin D. A., Borisenkova I. V. The influence of the dispersion of alumina slag and sulfoaluminate clinker on the formation of the structure of cement stone. Technique and technology of silicates. 2011. V. 18. No. 2. pp. 12-14.
15. Samchenko S. V., Zorin D. A. The effect of the dispersion of the expanding component on the properties of cements. Technique and technology of silicates. 2006. T. 13. No. 2. pp. 2-7.
16. Samchenko S. V., Zemsikova O. V., Kozlova I. V. The effect of the dispersion of the slag component on the properties of slag Portland cement. Technique and technology of silicates. 2016. Vol. 23. No. 2. pp. 19-23.
17. Kozlova I. V. The influence of ultrafine slag component on the properties and structure of cement stone. Successes of modern science. 2017. V. 5. No. 4. pp. 7-11.

#### Article 4

**Aung Htut Thu, Zakharov A.I.**

**The heat-insulating material on the silicate sheaf received on the basis of waste of rice processing**

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**Keywords:** thermal insulation, liquid glass, rice peel

#### Abstract

In this work use of waste of agriculture (a rice peel) as raw materials for production of heat-insulating materials is considered that provides decrease in energy consumption, increase in profitability and environmental friendliness of production.

#### References

1. Perspektivnyye metody pererabotki risovoy luzgi. [Elektronnyy resurs]. Rezhim dostupa: [http://www.newchemistry.ru/printletter.php?%20n\\_id=%206216](http://www.newchemistry.ru/printletter.php?%20n_id=%206216). Data obrashcheniya: 21.7.2018
2. S. Chandrasekhar., Review Processing, properties and applications of reactive silica from rice husk—an overview // K. G. Satyanarayana, P. N. Pramada, P. Raghavan, T. N. Gupta, Journal of Materials Science, August 2003, Volume 38, Issue 15, p. 3159 – 3168.
3. Bajirao S. Todkar<sup>1</sup>, Extraction of Silica from Rice Husk/ Onkar A. Deorukhkar, Satyajeet M. Deshmukh// International Journal of Engineering Research and Development, Volume 12, Issue 3 (March 2016), P.69 - 74.

4. Sun, L. Y.; Gong, K. C., Silicon-based materials from rice husks and their applications. *Industrial & Engineering Chemistry Research* 2001, 40 (25), P. 5861 - 5877.
5. Kim, H.-S.; Yang, H.-S.; Kim, H.-J.; Park, H.-J., Thermogravimetric analysis of rice husk flour filled thermoplastic polymer composites. *Journal of Thermal Analysis and Calorimetry* 2004, 76 (2), P. 395 - 404.
6. GOST R 54854-2011. Betony legkiye na organicheskikh zapolnitelyakh rastitel'nogo proiskhozhdeniya. Tekhnicheskiye usloviya [Elektronnyy resurs]. Rezhim dostupa: <http://docs.cntd.ru/document/gost-r-54854-2011>, (Data obrashcheniya: 21.9.2018)
7. Vysokoeffektivnyy penobeton s primeneniym zoly risovoy shelukhi. [Elektronnyy resurs]. Rezhim dostupa: <http://tekhnosfera.com/vysokoeffektivnyypenobeton-s-primeneniem-zoly-risovoy-sheluhi> (Data obrashcheniya: 22.7.2018)
8. Korneyev V.I., Danilov V.V. Zhidkoye i rastvorimoye steklo.- SPb.: Stroyizdat, 1996. — 216 s.
9. Tkhu A.KH., Zakharov A.I. Polucheniye neorganicheskoy svyazki dlya kholodnotverdeyushchikh smesey. *Novyye ogneupory*. 2018 №6. S.41 - 45.
10. Aung Khtut Tkhu, Zakharov A.I., Malyarov A.I. Svoystva kholodnotverdeyushchikh smesey s zhidkostekol'nym svyazuyushchim, poluchennym iz sel'skokhozyaystvennykh otkhodov. *Liteynoye proizvodstvo*, №9, 2018, S.22 - 25
11. Satta Panyakaew., Agricultural waste materials as thermal insulation for dwellings in Thailand: preliminary results/ Steve Fotios/ In: (Passive and Low Energy Architecture) PLEA-25th conference on passive and low energy architecture, Durbin, 22-24October 2008, P. 3-21.
12. Axel Berge, Pär Johansson., Literature Review of High Performance Thermal Insulation, Report in Building Physics, Gothenburg, Sweden 2012. [Elektronnyy resurs]. Rezhim dostupa: [http://publications.lib.chalmers.se/records/fulltext/local\\_159807.pdf](http://publications.lib.chalmers.se/records/fulltext/local_159807.pdf) (Data obrashcheniya: 21.9.2018)
13. Arbolitovyye bloki - nedostatki, dostoinstva i kharakteristiki [Elektronnyy resurs]. Rezhim dostupa: <https://srbu.ru/stroitelnye-materialy/223-arbolitovyebloki-nedostatki-dostoinstva-i-kharakteristiki.html> (Data obrashcheniya: 21.9.2018)
14. Arbolit Linii po proizvodstvu arbolita [Elektronnyy resurs]. Rezhim dostupa: <http://www.arbolit.com/harakteristiki#> (Data obrashcheniya: 21.9.2018)

## Article 5

**Vasilkov O. O., Barinova O. P., Kirsanova S. V., Elfimov A. B., Marnautov N. A.**

### **Influence of temperature on synthesis by spontaneous crystallization of nickel chromite**

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**Keywords:** nickel chromite, chromonickel spinel, infrared ranges, morphology of crystals, simple forms of an facet octahedron and tetragontrioctahedron

### **Abstract**

Influence of temperature on synthesis process by spontaneous crystallization of nickel chromite  $\text{NiCr}_2\text{O}_4$  with structure of spinel is investigated. Steady formation of chromonickel spinel in a temperature interval of 900–1050 °C is established. X-ray diffraction characteristics (parameters of an elementary cell, density), morphological features, particle size distribution of chromite of nickel are defined.

### **References**

1. Ivanov V. V., Ulyanov A. K., Shabelskaya N. P. Chromite ferrites of transition elements: synthesis, structure, properties / Moscow: Edition «Akademiya Yestestvoznaniya», 2013. – 69 p.
2. Shabelskaya N. P. The formation of chromite transition elements / N. P. Shabelskaya, I. N. Zakharchenko, R. O. Vasilyeva, A. K. Ulyanov // *International journal of applied and fundamental research*. – 2013. – № 6. – P. 48.
3. Zubekhin A. P., Talanov V. M., Shabelskaya N. P. Method for producing catalyst based on nickel-copper chromite // Patent of Russia № 2207905. 2008.
4. Zaychuk A. V., Belyi Ya. I. Ceramic pigments of the black-brown series on the basis of slag steel-making // *Vestnik NTU «KPI»*. – 2012. – (965), № 59. – P. 25-35.
5. Abramovich B. G. Intensification of heat exchange by radiation using coatings - M.: Energiya, 1977. – 256 p.
6. Budnikov P. P., Ginstling A. M. Reactions in solid mixtures веществ. Moscow: Stroyizdat, 1965. 475 p.
7. Singh R. K., Yadav A., Narayan A., Singh A. K., Verma L., Verma R. K. Thermal, structural and magnetic studies on chromite spinel synthesized using citrate precursor method and annealed at 450 and 650°C // *J. Therm. Anal. Calorim.* - 2011. - V. 107, № 1. - P. 197-204.
8. Timokhin N. N., Nersesyan M. D., Borovinskaya I. P. The mixture to obtain black pigment // Patent of Russia № 2029746. 2005.

9. JSPDS powder diffraction database.
10. Ormond B. F. Structure of inorganic substances / Gos. izd-vo tekhniko-teoreticheskoy literatury. Moscow, 1950. - 968 p.
11. Narai S. Inorganic Crystal Chemistry. Budapest, 1969. 396-397 p.
12. Ptak M., Maczka M., Gagor A., Pikul A., Macalik L., Hanuza J. Temperatures-dependent XRD, IR, magnetic, SEM and TEM studies of Jahn-Teller distorted NiCr<sub>2</sub>O<sub>4</sub> powders // Journal of Solid State Chemistry - 2013 V. 201. - P. 270-279.
13. Ishibashi H., Yasumi T. Crystal structure of ferromagnetic phase of spinel compound NiCr<sub>2</sub>O<sub>4</sub>. // Photon Factory Activity Report. - 2006. - V. 23 (B). - P. 134.
14. Barros<sup>1</sup> B. S., A.C. F. M. Costa, R. H. A. G. Kiminami, L. da Gama. Preparation and characterization of spinel MCr<sub>2</sub>O<sub>4</sub> (M = Zn, Co, Cu and Ni) by combustion reaction // J. Met. & Nan.Mat. - 2004. - V. 20-21. - P. 325332.
15. Mosienko S. A. Coating material with high emissivity // Patent of Russia № 2262552. 2005.